

Class C & Class F: How Relevant are They for Predicting Performance?

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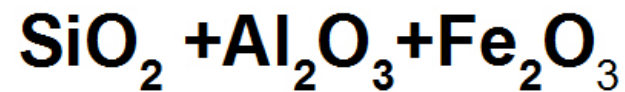
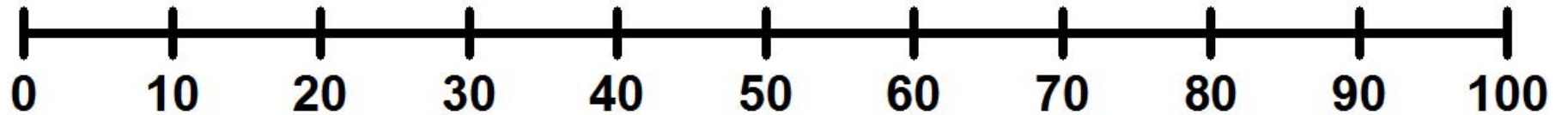
Outline

- ASTM C-618
- Fly ash Particle Characterization by ASEM
- Comparison with Class C & F
- Application of Cluster Analysis
- Development of a Performance-Based Reactivity Classification System
- Conclusions

ASTM C- 618

Class F

Class C



ISSUES

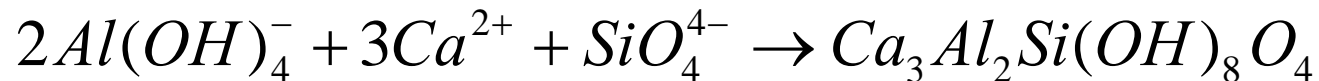
- Selection of oxides
- Selection of limits
- Significance of classes

$\text{SiO}_2 + \text{Al}_2\text{O}_3 + \text{Fe}_2\text{O}_3$ Reactions

Pozzolanic reaction



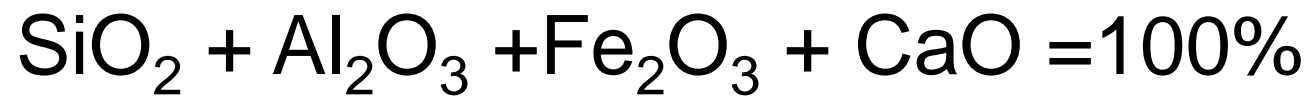
Hydrogarnet Formation



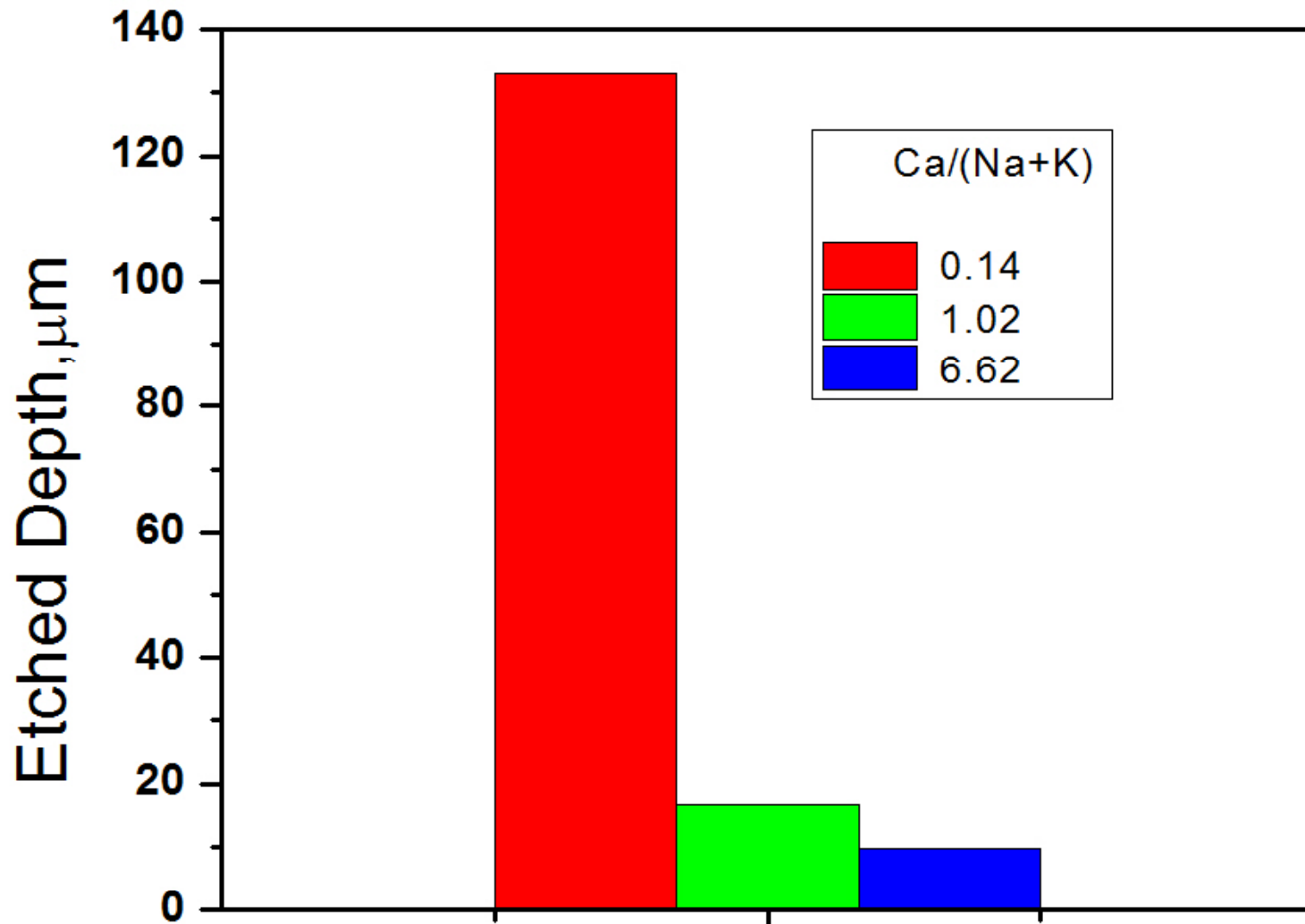
Limonite Formation



Calcium Effect



Silica Dissolution, Synthetic Fly Ash Glass in Simulated Porewater, 72 hr



ASTM C-618

- Class F = Bituminous coal fly ash
- Class C = Sub-bituminous coal fly ash

ASTM C-618

- Class F = ~~Bituminous coal fly ash~~
- Class C = ~~Sub bituminous coal fly ash~~

Bulk XRF Analysis

- Includes both reactive and inert phases
- Does not provide particle size data
- No information on variance of measured values

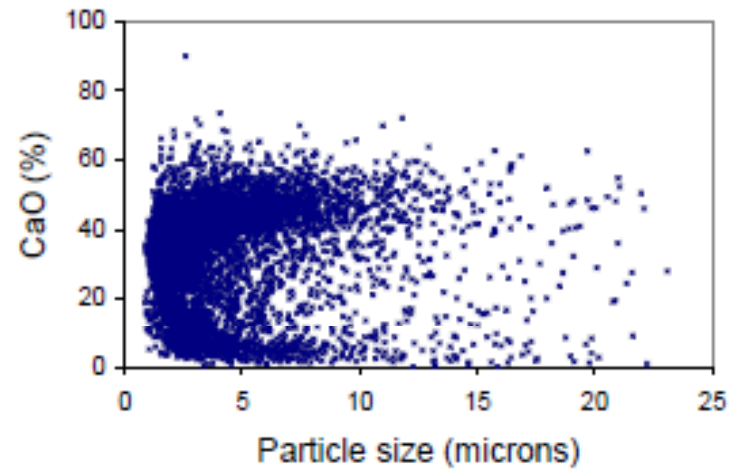
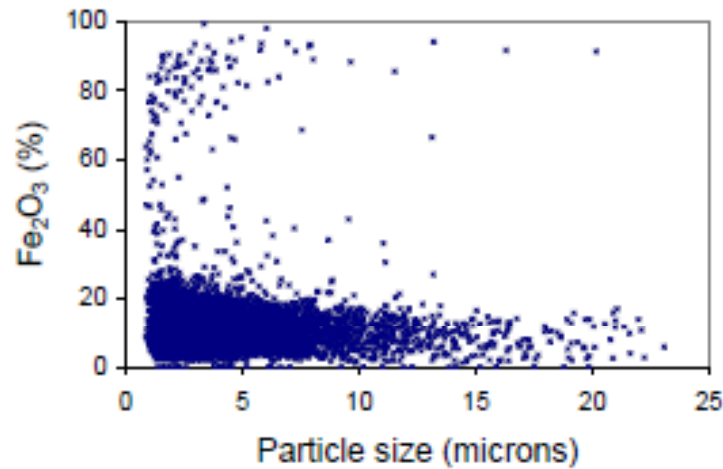
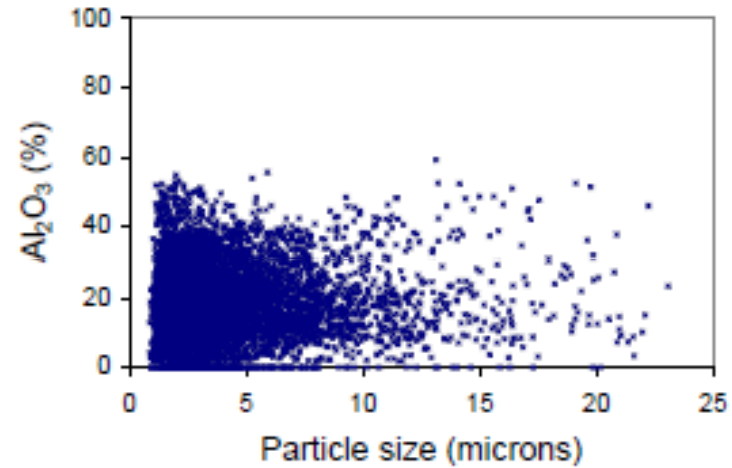
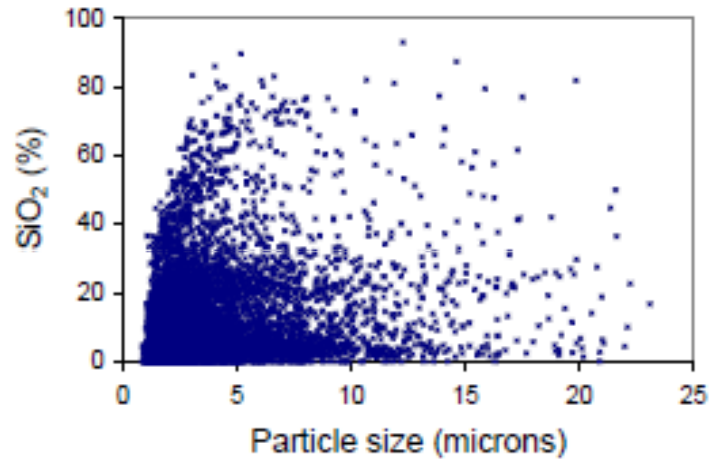
CCSEM (or PSEM or ASEM)

- SEM with Energy Dispersive XRF
- Computer-controlled stage
- Particle analysis software

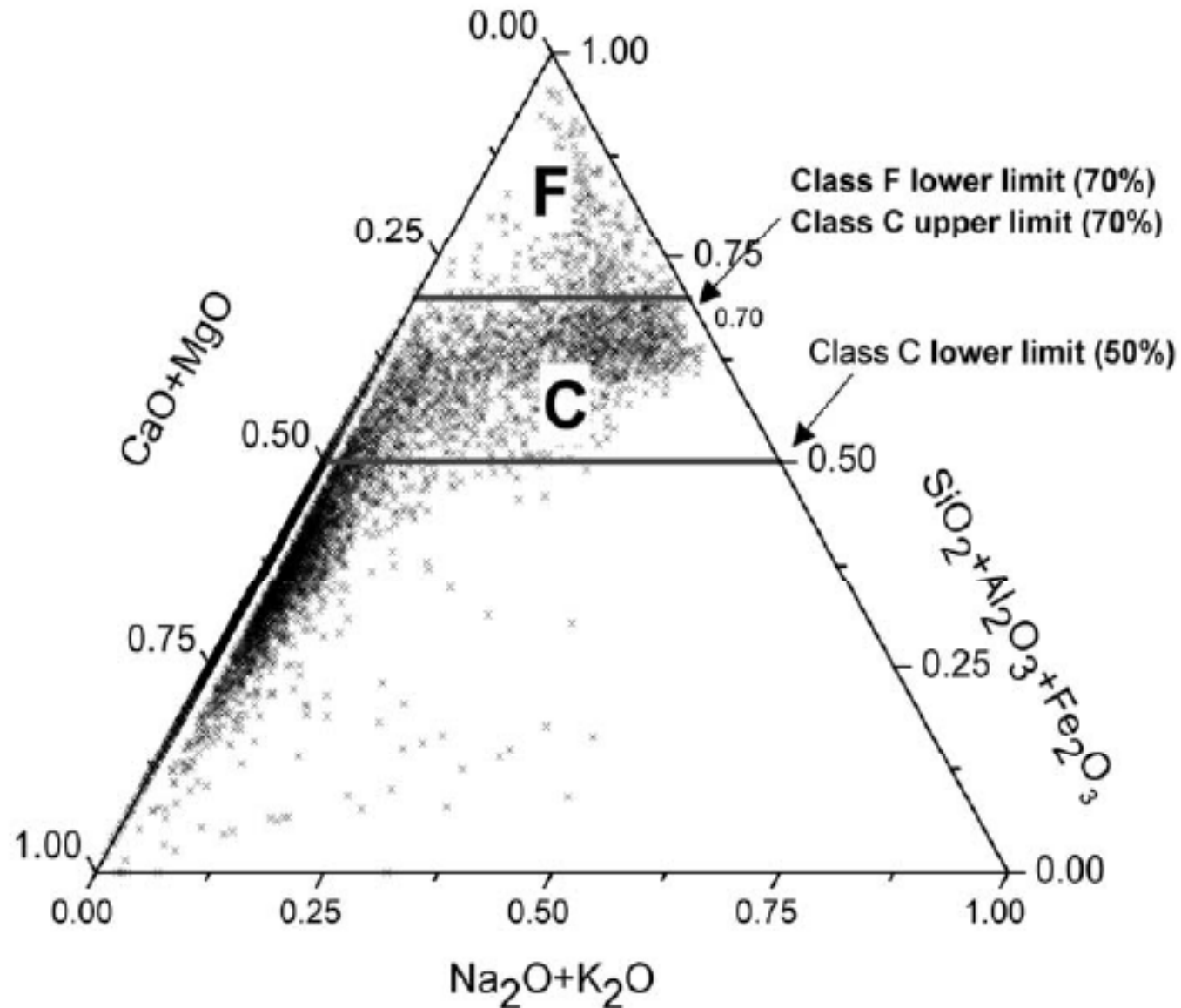
ASEM Data

- ~ 10,000 particles per sample
- 16 elements: Na, Mg, Al, Si, P, S, K, Ca, Fe, Ti, Fe, Ni, Zr, Ba, Ce, and Pb
- Glassy particles identified by:
 - Circularity - aspect ratio < 1.3
 - Size range 0.20 -25 microns

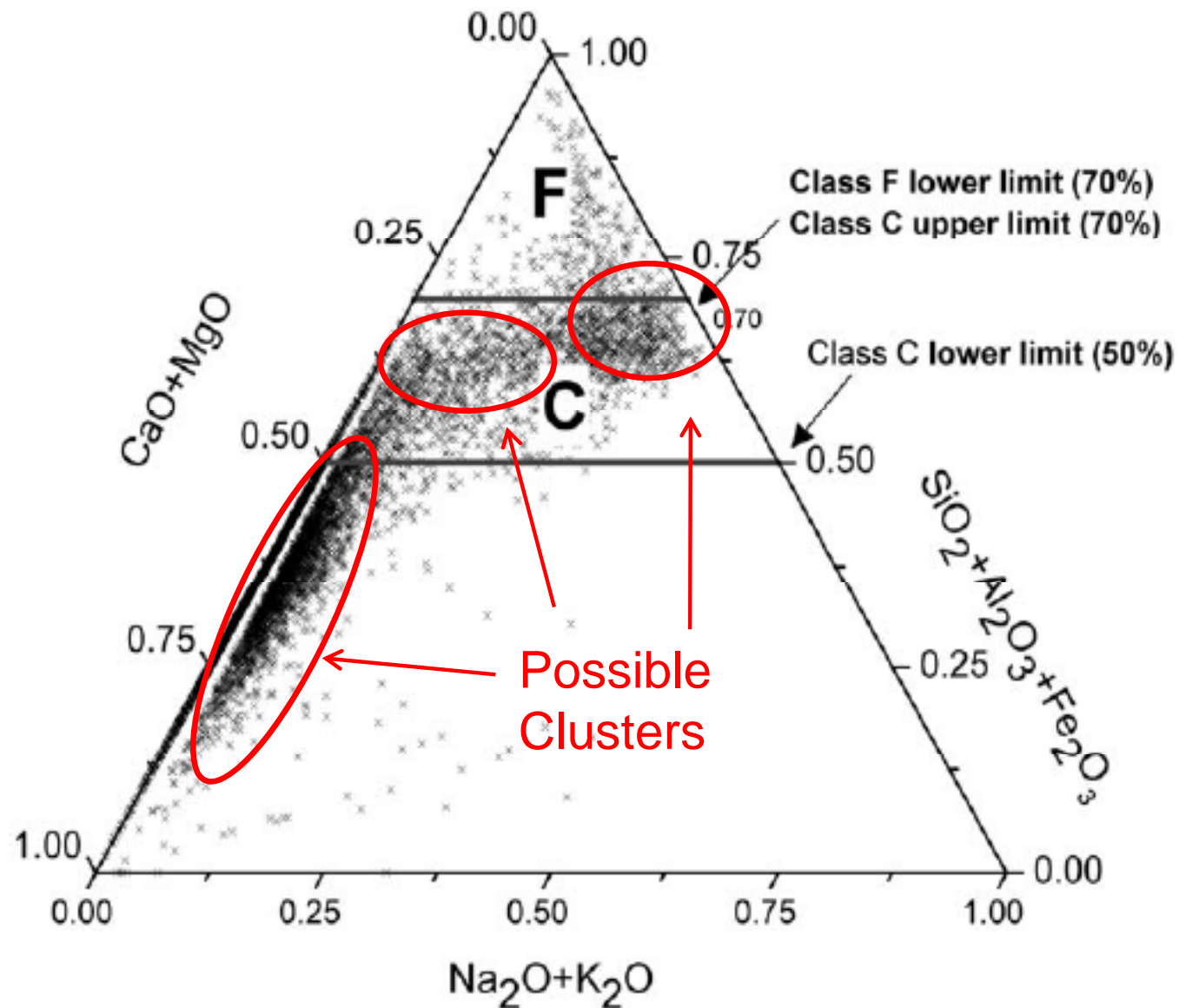
Element vs Particle Size



Coal Creek Fly Ash Particle Analysis



Coal Creek Fly Ash Particle Analysis

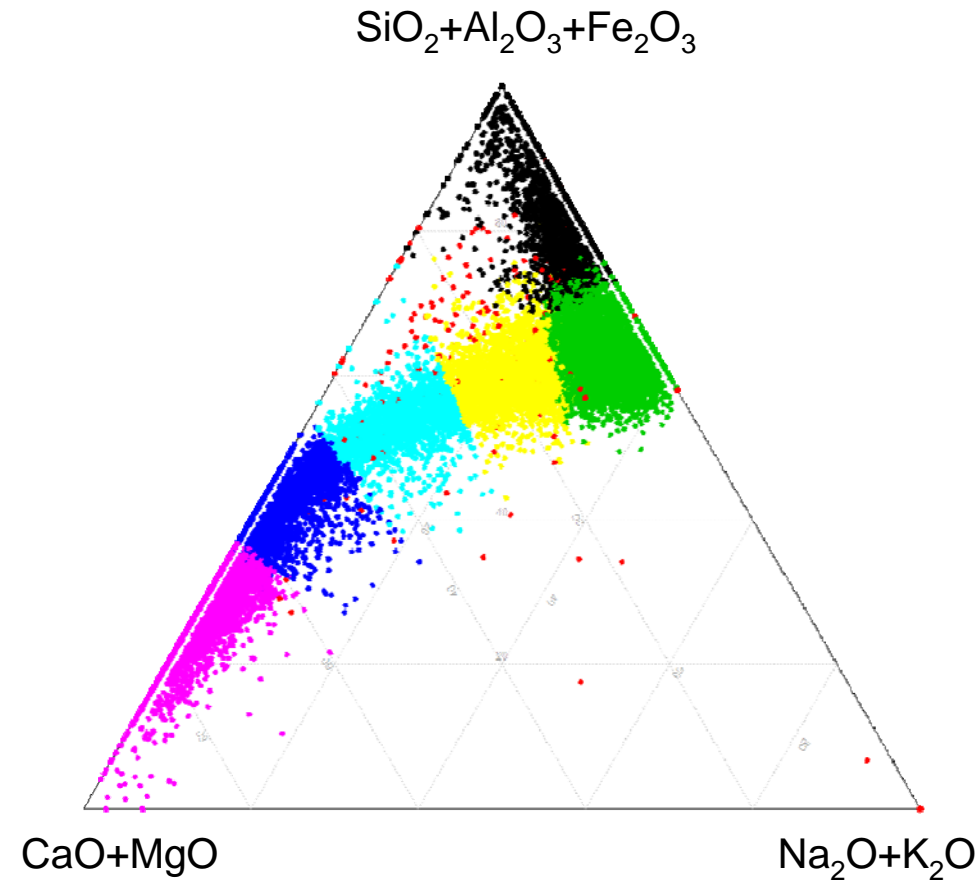


K-means algorithm

- 1) Pick a number (k) of cluster centers
- 2) Assign every data point to its nearest cluster center based on Euclidian distance
- 3) Move each cluster center to the mean of its assigned data point
- 4) Repeat 2-3 until convergence
- 5) Calculate the fraction of each cluster

Clusters of Coal Creek fly ash particles in glass coordinates

Table 1 Classification of Glassy Phase based on
Cluster Analysis of Coal Creek Fly Ash



Cluster No.	Color	Name	Number (%)
1	Black	Si-Al-Fe glass	9.57
2	Red	Trace element	1.00
3	Green	Si-Al-Fe glass	32.67
4	Blue	CAS glass	11.85
5	Cyan	CAS glass	11.93
6	Magenta	Ca rich glass	15.64
7	Yellow	CAS glass	17.34

Proposed Application of Clustered Fly Ash Composition

- Define a set of standard clusters based on statistical analysis of clusters analyzed from variety of actual fly ashes.
- The centroid of each of these standard clusters would be a specific chemical composition defining a standard glass.
- A given fly ash particle data set could then be classified in terms of this set of standard glasses
- The reactivity of each standard glass would be known

Development of a Performance-based Classification System

1. For a total number, N , of glassy particles analyzed by ASEM, each particle would be assigned to a standard glass class, based on its chemical composition. Then for k number of standard glass classes:

$$N = \sum_{i=0}^k n_i$$

where n_i is the number of particles in the i^{th} class

Approach (con't)

2. To convert them to mass fractions it is necessary to use the individual particle radius and the density of the glass.

$$x_i = \frac{\sum_{j=1}^{n_i} r_j^3 \rho_i}{\sum_{i=1}^k \sum_{j=1}^{n_i} r_j^3 \rho_i}$$

where x_i is the mass fraction, r_j is the radius of the j^{th} particle and ρ_i is the density of the i^{th} standard glass

Conclusions

- ASTM C-618 Classification system does not include all the chemically relevant oxides
- Rationale for Class C and Class F limits is not clear
- ASTM C-618 does not provide a reliable prediction of reactivity
- ASEM provides data on individual fly ash particles and makes it possible to discriminate between reactive and inert phases
- Particle-based data make it possible to identify characteristic clusters of glass particles with similar compositions.
- These clusters can then be used to develop a standard set of fly ash glass compositions.



Thank you for your attention

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