

Carbon Capture Technologies and Implementation Challenges for the Cement Industry

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RTI International

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RTI International

- The second largest non-profit research institute in the US
- Mission: To improve the human condition by turning knowledge into practice
- Headquartered on 180-acre campus in Research Triangle Park, N.C.
- Founded in 1958
- FY2008 Revenue - \$ 710 million
- 2600 staff working in over 40 countries



CO₂ in the Cement Industry

- Cement industry releases 1.3 – 1.7 GtCO₂ annually
 - ~5% of global anthropogenic CO₂ emissions
 - General metric: 1 ton CO₂ emitted for every 1 ton clinker produced
- Two sources of CO₂:
 - *Raw material-derived*: $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$
 - *Fuel-derived*: $\text{C} + \text{O}_2 \rightarrow \text{CO}_2$
- What CO₂ regulatory environment awaits us? Who knows?
 - Multiple Congressional bills proposed (e.g. Lieberman-Warner, etc.); AB 32 legislation in California; Europe's cap and trade
 - Some form of carbon regulations are probably on the horizon
 - Primary target: Fossil-fuel fired power plants (~43% of global CO₂ emissions)
 - Secondary targets: Cement plants, steel mills, refineries

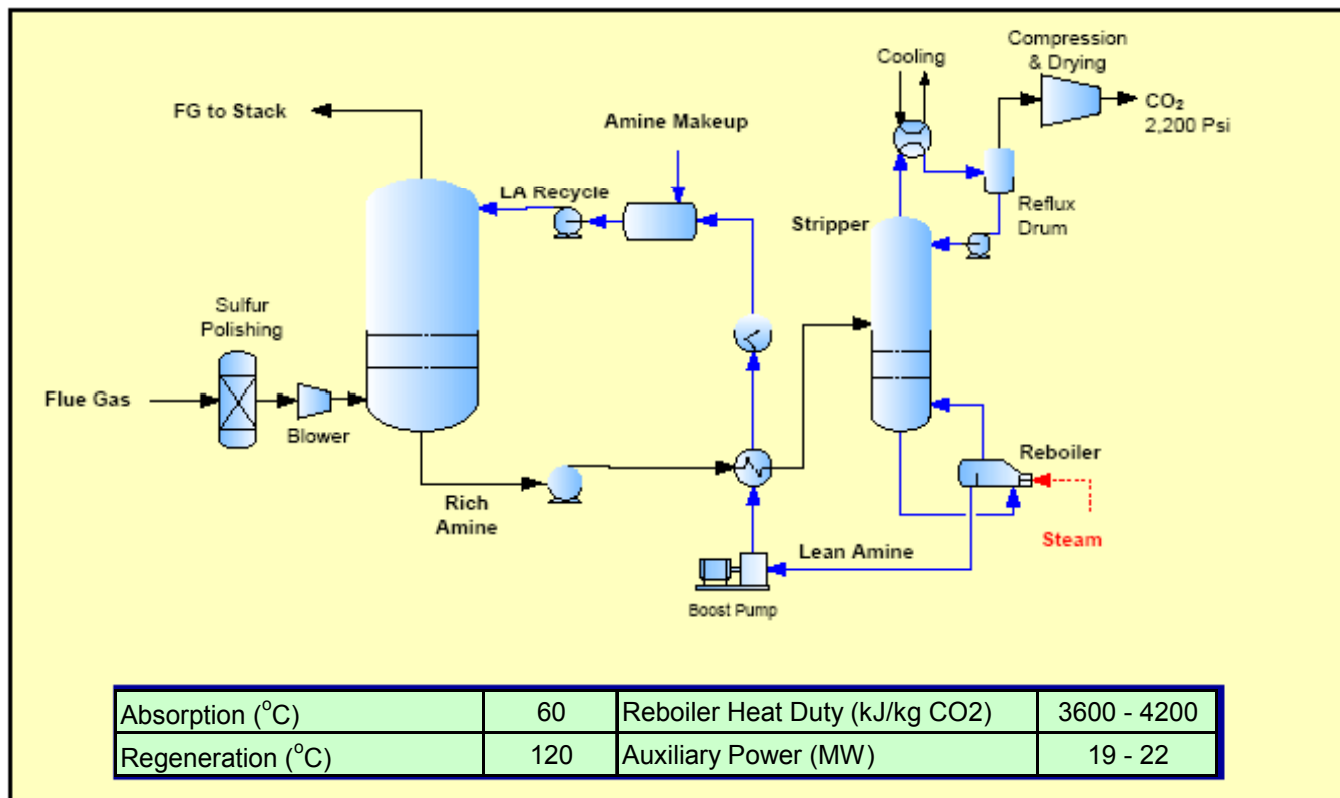
Carbon Mitigation Options

Carbon mitigation options

- Energy efficiency, blended cements, alternative binders, carbon neutral fuels, and...
- ...several post-combustion CO₂ capture technologies are compatible with cement plant exhausts
 - Chemical solvents: amines, ammonia, carbonate solvents
 - Physical solvents: methanol, ionic liquids
 - Membranes: polymers, molecular sieves
 - Physical sorbents: metal organic frameworks
 - Chemical sorbents: carbonates, CaO, amine-enhanced zeolites

Conventional Technology: Monoethanolamine (MEA)

Amine CO₂ scrubbing



MEA Process

- Proven
- Effective

but...

- Energy intensive
- Expensive
- Corrosive
- Hazardous waste produced

Source: *Existing Coal Power Plants and Climate Change...*, 2008 Energy Technology Forum, Ciferno J., DOE/NETL

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RTI's Dry Carbonate Process

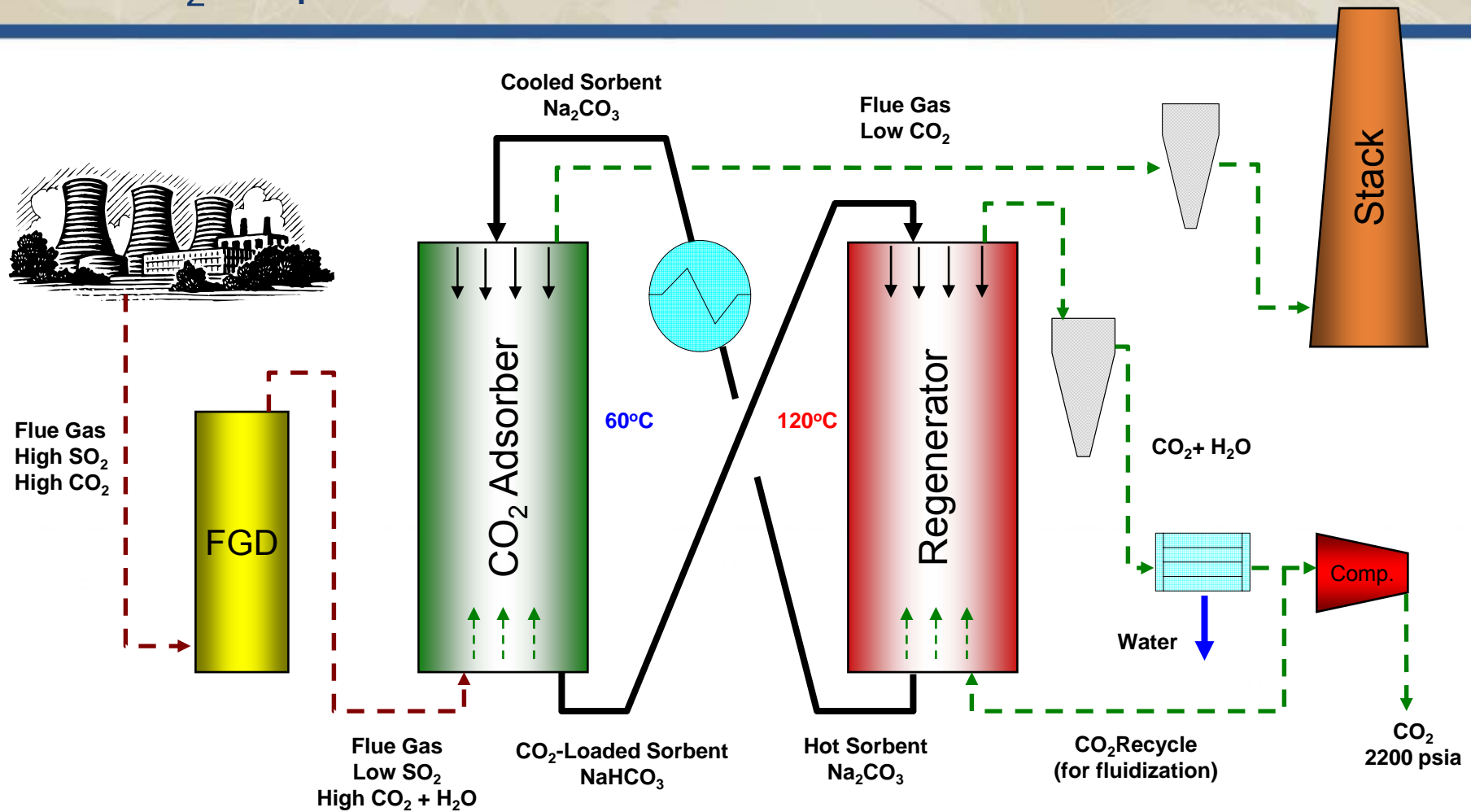
RTI's objective has been to develop a carbon dioxide capture technology that is

- Based on a solid, regenerable, carbonate sorbent
- Applicable to flue gases of coal and natural gas-fired power plants
- Intended for retrofit in existing plants
- Less expensive and less energy intensive than conventional technologies
- Of relatively simple process design

7 years of R&D effort with \$7MM of DoE funding committed to date.

The Dry Carbonate Process

CO₂ Capture from Flue Gas



The Dry Carbonate Process

A Closer Look at Reaction Chemistry

CO₂ Adsorption (Carbonation)



Exothermic $\Delta H_r^\circ = -3082 \text{ kJ/kg CO}_2$

Operating temperature: < 80°C

Sorbent Regeneration



Endothermic $\Delta H_r^\circ = 3082 \text{ kJ/kg CO}_2$

Operating temperature: > 100°C

Contaminants



Reactions with SO₂ and HCl are irreversible at process conditions

No observed effects by O₂, Hg, and NO_x

Advantages and Challenges of Using Sodium Carbonate

Advantages

- Lower total regeneration energy requirement (vs. conventional MEA)
- Lower CO₂ removal cost
 - Readily available, abundant, and inexpensive sorbent.
 - No flue gas pretreatment (no heating, no cooling, no guard beds)
- Non-hazardous and non-toxic sorbents; no hazardous waste generated

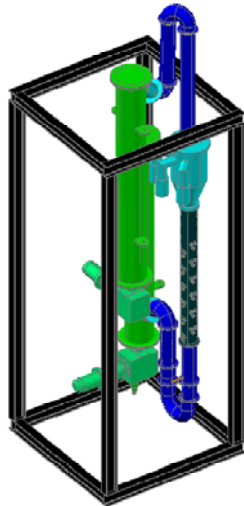
Challenges

- Large solids handling/circulation requirements
 - Best-case scenario → Na₂CO₃ : CO₂ = 2.4 : 1 (mass ratio)
 - *High capacity sorbents; best available solids handling technologies*
- Exothermic CO₂ adsorption affects reaction equilibrium
- CO₂ removal requires equimolar amount of water
- Condensed water causes raw Na₂CO₃ to agglomerate

RTI's Technology Development

Sorbent Development

- Raw materials
 - *Higher theoretical capacity (~40%), inexpensive*
- Supported sorbents
 - *Improved reactivity, improved attrition resistance, improved ability to adsorb heat during exothermic CO₂ removal*
- ~225 kg supported sorbent manufactured to date



Process Development

- Entrained-bed reactor (Dispersed phase)
 - *> 90% capture achieved*
 - *Better heat control, better sustained reactivity*

The Dry Carbonate Bench-Scale Unit



System Specifications

- Designed to “treat” up to 200 SLPM of flue gas
- Sorbent circulation rate: 10 – 150 kg/hr
- Screw conveyors: 20 cm wide and 2 m tall
- Heated screw conveyor is rated to 550 kPa (~160°C saturated steam)
- City water used for cooling

Note:

This “integrated” system was built after first verifying the performance of each process component individually

Field Testing of the Dry Carbonate Unit

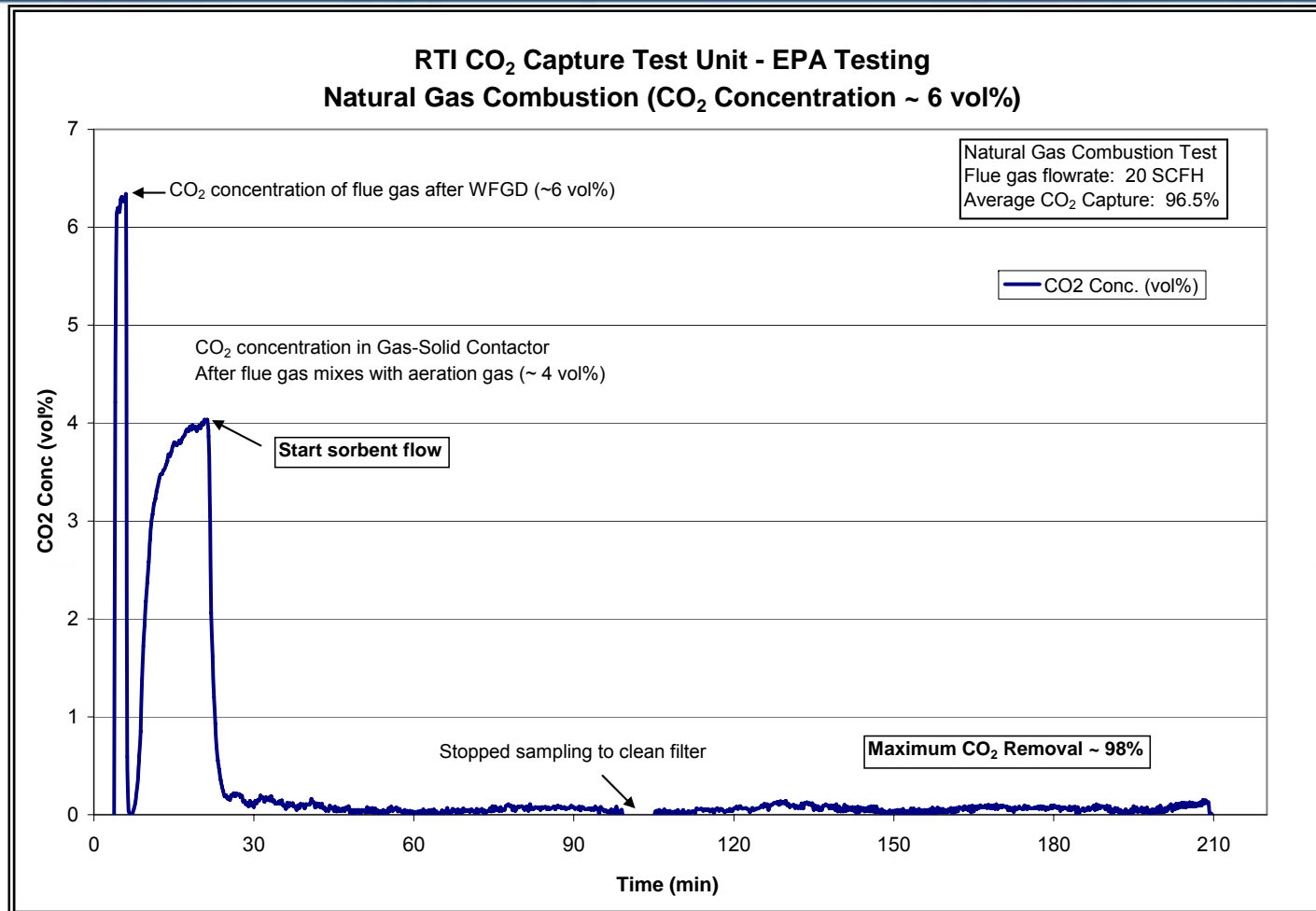
U.S. EPA – Research Triangle Park, NC

- EPA's Multi-pollutant Control Research Facility
 - 4.2 Million kJ/hr (1.2 MW_t) multi-fuel fired facility
 - ~10 tonne/day CO₂ produced
 - Field test site used to evaluate emission control technologies
- RTI's Dry Carbonate Unit installed and operated at EPA site
 - Testing was coordinated with ARCADIS, Inc. (EPA's on-site contractor)
 - ~3-5% slipstream of EPA's flue gas
- Objectives
 - Evaluate process and sorbent using actual combustion flue gas
 - Identify optimal operating conditions
 - Evaluate sorbent degradation and effects of flue gas contaminants



RTI's Dry Carbonate Process prototype was tested at EPA for >230 hrs using coal & natural gas flue gas

Field Testing at U.S. EPA Natural Gas Combustion



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Key Findings of Field Testing at U.S. EPA

- Sorbent is capable of sustained >90% CO₂ capture in both coal- and natural gas-derived flue gas
- Sorbent regeneration more effective at 120 - 140°C (EPA steam header)
- No negative effects observed due to contaminants
 - No significant particle changes, no agglomeration, CO₂ capture not inhibited
- After > 1000 hrs of circulation, sorbent shows minor signs of wear
 - Slightly lower avg. particle size, surface area
 - Larger scale testing will be better for attrition analyses
- Further work required to optimize:
 - heat control in the adsorption unit to improve heat removal from the sorbent
 - Sorbent loading capacity for economical operation of process

Current R&D Findings and Path Forward

Bench-scale findings

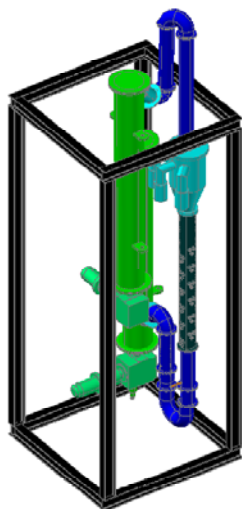
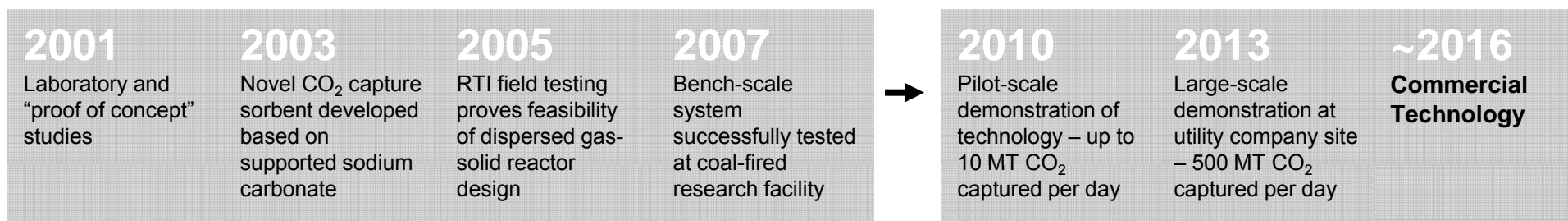
- Sorbents with improved CO₂ loading capacities produced [> 15 wt% CO₂]
- Modified designs for the adsorption and regeneration reactors have been evaluated
- CO₂ capture and loading rate are greatly improved with improved heat transfer in absorber

Process development and demonstration

- Construction of a 0.3 MT CO₂ captured per day unit
- Long-term testing planned at RTI's Central Utility Plant and EPA site
- Testing expected to begin in mid-2009, last 8-10 months
- Data will be used to scale-up to 5 - 10 MT CO₂ capture per day
- Perform a comprehensive economic analysis

Path to Commercialization

Anticipated development timeline for RTI's Dry Carbonate Process



Cement Plant vs. Power Plant

Comparison of exhaust gases at “typical” cement plant & power plant

	Cement Plant¹	PC Power Plant²
CO ₂ (vol%)	22.4	13.1
H ₂ O (vol%)	7.2	17.3
SO ₂ (vol%)	n/a	< 0.01
O ₂ (vol%)	2.3	2.4
N ₂ (vol%)	68.1	66.4
Temperature	160°C	60°C
Pressure	1 atm	1 atm
Flow rate	3,043 m ³ /min	44,886 m ³ /min
CO ₂ Flow rate	~81 MT/hr	~765 MT/hr
Solids feed rate	~100 MT/hr (kiln feed)	~295 MT/hr (coal feed)

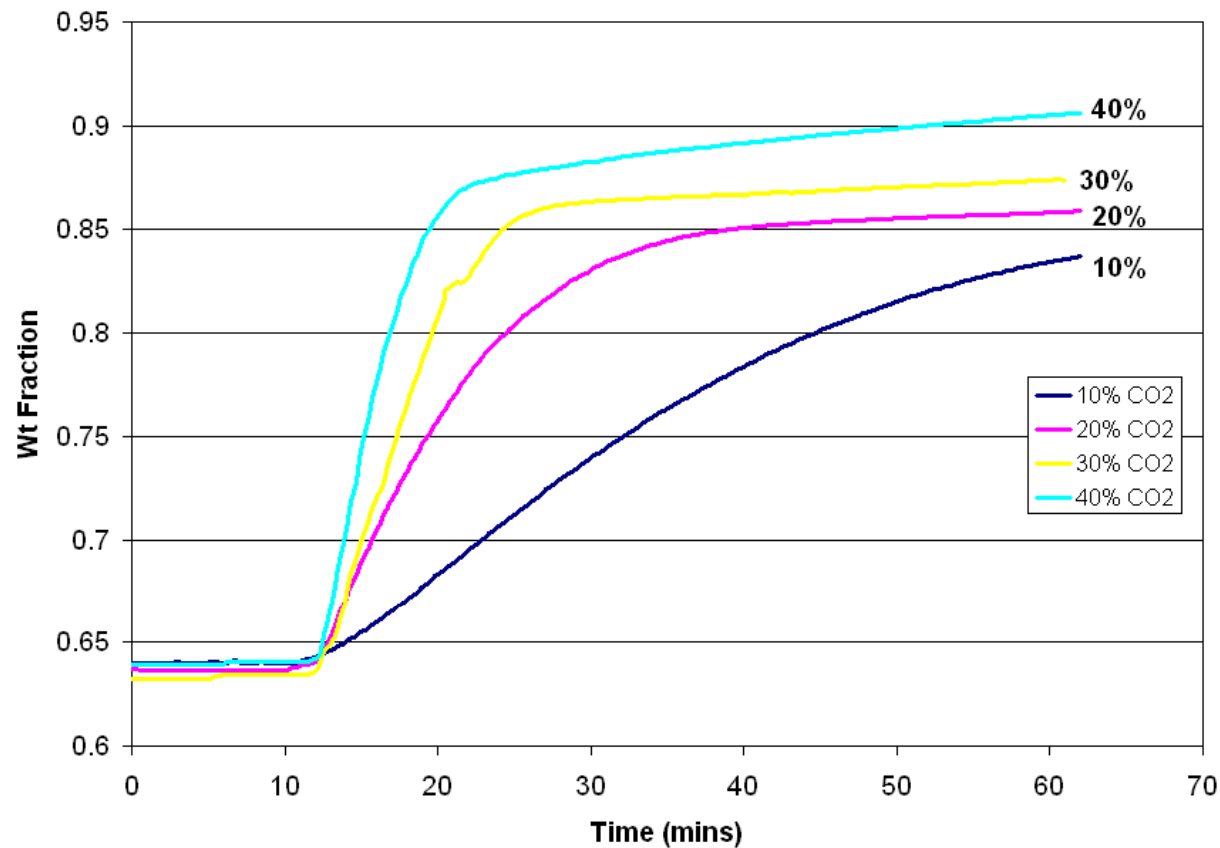
¹Based on St Marys cement plant case study in S. M. Nazmul Hassan thesis, University of Waterloo, 2005 (2,400 TPD clinker rate)

²Based on case study in DOE's *Cost and Performance Baseline for Fossil Energy Plants*, August 2007 – 680 MWe PC power plant

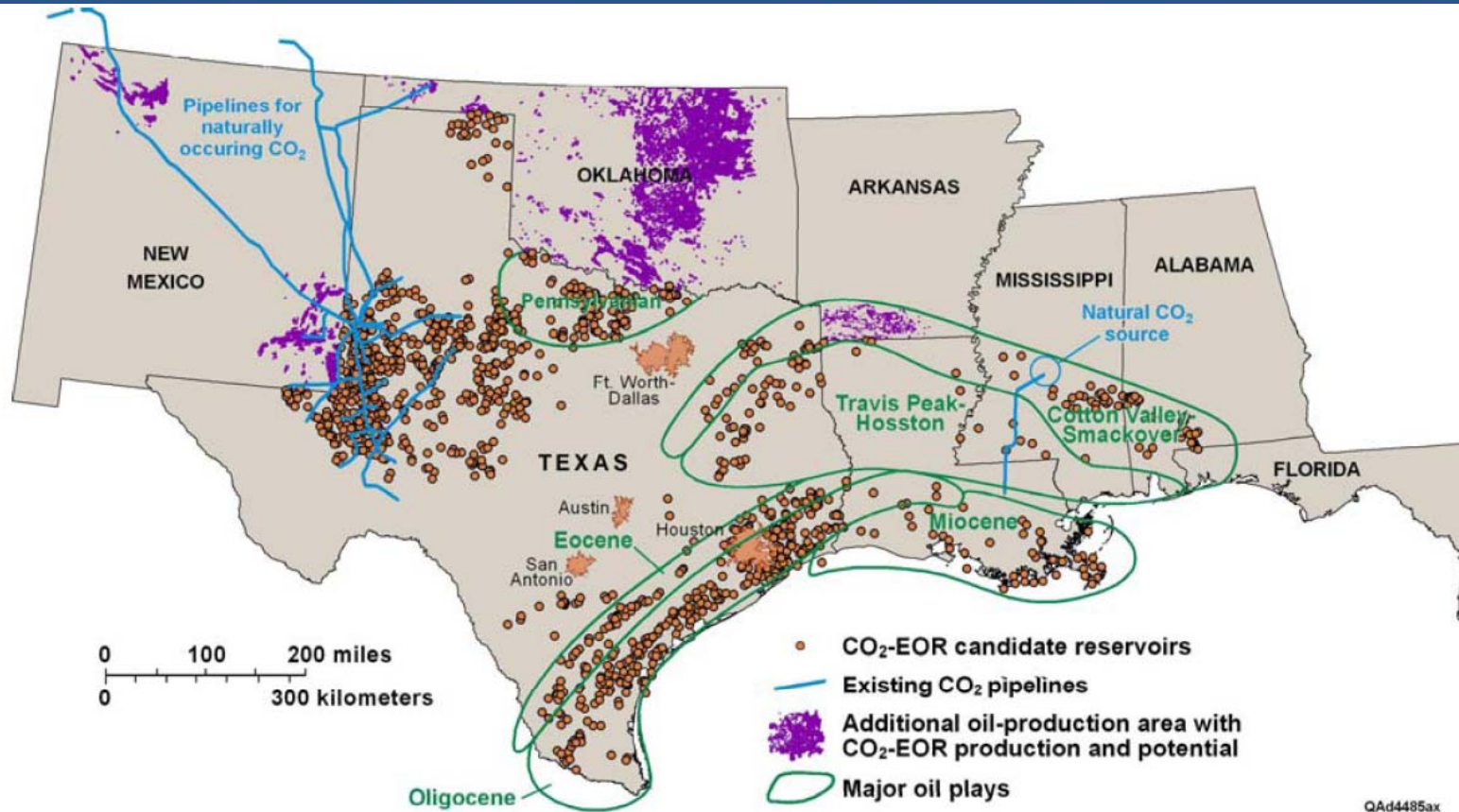
Higher CO₂ Concentration

Comparison of CO₂ capture rates at various CO₂ feed concentrations

TGA Data



CO₂ EOR Potential in the Gulf Coast



Ref: West Texas Geological Society Publ. #05-115
Fall Symposium, October 26-27, 2005
Mark H. Holtz, Vanessa Nunez Lopez, and Caroline L. Breton

Continuation of Cement Plant Evaluations

- Comprehensive economic analysis of Dry Carbonate Process installed at a cement plant
 - Capital & operating costs based on \$/kg CO₂ removed & \$/ton cement produced
- Evaluation of equipment size, full integration scheme, materials handling & storage, land use, resource availability, etc
- Testing of new sorbents under high CO₂ concentration and simulated cement exhaust gas
- Evaluation of CO₂ compression to pressures needed for pipeline delivery of CO₂ for EOR applications